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# Synthesis of GA<sub>20</sub> Glucosyl Derivatives and the Biological Activity of Some Gibberellin Conjugates

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Abstract.  $GA_{20}$ -13-0-glucoside (7*a*) and  $GA_{20}$  glucosyl ester (6*a*), potential endogenous conjugates in maize, were synthesized chemically. The biological activities of these compounds and of nine more GA glucosyl derivatives were determined using the Zea mays dwarf-5 seedling and Oryza sativa cv. "Tan-ginbozu" assays. The relative bioactivities of the conjugates were also calculated.

Gibberellins (GAs) are known to be involved in the regulation of many physiological processes in plants. Qualitative as well as quantitative changes of the GA content are assumed to parallel physiological effects. Thus, both identification and quantification of all GAs in a plant are necessary in order to study the mechanism of regulation (Metzger and Zeevaart 1980, Hedden et al. 1982, Yamaguchi et al. 1982). While recent studies have taken into account the kinds and amounts of free GAs, GA conjugates have received little attention, even though they may sometimes represent the major endogenous GA content or represent the main products from feeding experiments (e.g., Yamane et al. 1975, Rood et al. 1982). This situation is the result of difficulties in analyzing trace amounts of highly polar substances—e.g., GA glucosyl derivatives. Also, the physical and biological properties of only a few of the derivatives are known. A cautious interpretation has to be made of the released free GAs that come from the enzymatic or chemical hydrolysis of GA conjugates (Schneider 1983).

We became interested in GA conjugation in maize, and especially  $GA_{20}$ ,

Abbreviations: GA, gibberellin; <sup>1</sup>H NMR, proton nuclear magnetic resonance; TMS, tetramethylsilane; HPLC, high-performance liquid chromatography.

because of recent studies on the content and metabolism of GAs in Zea mays (Hedden et al. 1982, Heupel et al. 1982, Wurtele et al. 1982, Phinney and Spray 1982).  $GA_{20}$  is the gibberellin immediately prior to  $GA_1$ , and  $GA_1$  is probably the only active gibberellin per se in maize (Phinney and Spray 1982). Channeling  $GA_{20}$  into a conjugate could be an important way of regulating the level of free  $GA_1$ . In this study we report the synthesis of two  $GA_{20}$  glucosyl derivatives together with information on the bioactivities of nine GA glucosyl derivatives.

#### **Materials and Methods**

### Synthesis of GA<sub>20</sub>-13-0-glucoside (7a)

Gibberellin  $A_{20}$  methyl ester (5b) (560 mg, 1.6 mmole) (Beale et al. 1980) and  $Ag_2CO_3$ /celite (5.8 g, ca. 10 mmole) in dichloroethane (30 ml) were reacted for 15 min with  $\alpha$ -acetobromoglucose (3.5 g, 8.0 mmole) dissolved in dichloroethane (2 ml) under reflux (Schneider 1981a,b). The filtered and evaporated mixture was deacetylated by 25 ml of 0.5 N sodium methoxide (1 h at room temperature). Subsequent column chromatography on 60 g silica eluted with chloroform containing 0, 5, 10, 15, 20, 25% methanol (150 ml each concentration, 50-ml fractions) gave unreacted GA<sub>20</sub> methyl ester (5b) (220 mg = 58.9% yield) in fractions 4–8 and GA<sub>20</sub>-13-0-glucoside methyl ester (7b) (322 mg = 39.8% yield) in fractions 12–15.

The methyl ester of  $GA_{20}$ -13-0-glucoside (7b) (300 mg, 0.59 mmole) was treated with lithium propyl thiolate (3.2 mmole) in hexamethyl phosphoramide under nitrogen for 3 h. Column chromatography on 30 g silica (chloroform:methanol:acetic acid; 50:0:0, 45:5:0, 40:10:1, 40:10:2, 40:10:4, 15 ml-fractions) yielded 226 mg amorphous  $GA_{20}$ -13-0-glucoside (7a) in fractions 11–14 (77.5% yield, total yield 30.9%). <sup>1</sup>H NMR (acetone-D<sub>6</sub>/TMS): 1.03 (s, 18-H<sub>3</sub>), 2.589 (m, 5- and 6-H), 4.553 (d, J = 7.7 Hz, 1-H), 4.953 and 5.322 ppm (m, 17-H<sub>2</sub>).

#### Synthesis of GA<sub>20</sub> glucosyl ester (6a)

Gibberellin  $A_{20}(5a)$  (50 mg, 0.15 mmol) (Beale et al. 1980) and  $Ag_2CO_3$ /celite (95 mg, ca. 0.17 mmole) in dichloroethane (4 ml) was stirred and heated. A solution of  $\alpha$ -acetobromoglucose (65 mg, 0.16 mmole) in dichloroethane (1 ml) was added. After 10 min the mixture was filtered, the solvent evaporated, and the residue, dissolved in methanol, applied to a 15-ml DEAE-Sephadex A-25 column. The column was eluted stepwise with 15-ml aliquots of methanol, 0.5 N acetic acid-methanol, 1.0 N acetic acid/methanol, 3.0 N acetic acid/methanol; 5 ml fractions were collected.

Fractions 3-5 contained a neutral mixture, which was rechromatographed on a column of 8 g silica with petroleum ether/ethyl acetate, giving 48 mg GA<sub>20</sub>- $\beta$ -D-2,3,4,6-tetra-0-acetyl-glucosyl ester (6b) (49% yield); crystals from methanol, m.p. 142-144° decomp. Fractions 13-14 contained 25 mg (48%) of un-

#### GA Glucosyl Derivatives

reacted GA<sub>20</sub> (5*a*); (6*b*) (28 mg, 0.04 mmole) in methanol (1 ml) was reacted with 0.5 N sodium methoxide (30  $\mu$ l) for 5 min at room temperature. The crude mixture was separated on TLC (silica; chloroform:methanol:water:acetic acid; 80:30:2:5). Elution of the zone of R<sub>f</sub> = 0.8–0.6 with methanol, followed by redissolving in acetone, gave 12 mg of GA<sub>20</sub>-β-D-glucosyl ester (6*a*) (57%; total yield = 28.0%). <sup>1</sup>H NMR (acetone-D<sub>6</sub>/TMS): 1.027 (s, 18-H<sub>3</sub>), 2.608 (d, J = 10.26 Hz, 6-H), 2.689 (d, J = 10.26 Hz, 5-H), 4.850 and 5.174 (m, 17-H<sub>2</sub>), 5.535 ppm (d, J = 7.72 Hz, 1-H).

## HPLC Conditions

A Serva Si-polyol C-18 column (4.6  $\times$  250 mm) was fitted with an RCT HPLC eluent supply and a Pye/Philips PU 4020 detector set on 206 nm. Isocratic elution with methanol:0.1% phosphoric acid (1 ml/min) was used; 2-5 µg of each compound per injection was applied.

## Dwarf-5 Maize Assay

The experimental details of the *dwarf-5* maize assay used in these studies have been described previously (Phinney and Spray 1982).

# Dwarf Rice Assay (cv. "Tan-ginbozu")

The bioassay was performed according to Sembdner et al. (1976). Substances were applied via the roots.

## **Results and Discussion**

Gibberellin  $A_{20}$  (5a) and  $GA_{20}$  methyl ester (5b) (Fig. 1) were reacted with  $\alpha$ acetobromoglucose in the presence of silver carbonate/celite (Schneider 1981a,b). In the case of  $GA_{20}$  methyl ester (5b) the 13-hydroxyl group was glucosylated with a yield of 39.8%, resulting in  $GA_{20}$ -13-0-glucoside methyl ester (7b). Demethylation of (7b) gave  $GA_{20}$ -13-0-glucoside (7a) with a total yield of 30.9%. The doublet at 4.553 ppm (J = 7.7 Hz) in the <sup>1</sup>H NMR spectrum confirms the  $\beta$ -D-glucopyranoside structure.

The reaction of equivalent amounts of free GA<sub>20</sub> (5*a*) and  $\alpha$ -acetobromoglucose under the same conditions led to the ester tetraacetate (*6b*) (m.p. 142– 144°), which by short-term deacetylation gave GA<sub>20</sub> glucosyl ester (*6a*) (28% total yield), showing the characteristic 1-H doublet at 5.535 ppm (J = 7.72 Hz) for the glucosyl ester linkage in the <sup>1</sup>H NMR. The purity of the synthesized conjugates was checked by reverse-phase HPLC (Table 1) (Schneider 1983, Koshioka et al. 1983). The results show that reverse-phase HPLC is capable of discriminating glucosides and glucosyl esters as well as other interfering conjugates. However, a comparison of the data reveals that the chromatographic behavior of unknown conjugates can rarely to predicted. For example,



**Fig. 1.** Log:normal dose:response curves of  $GA_{20}(5a)$ —O—,  $GA_{20}$  glucosyl ester (6a)—x—, and  $GA_{20}$ -13-0-glucoside (7a)—**•**—of the *dwarf*-5 assay and calculations of the "relative activities."

 $GA_1$ ,  $GA_5$ , and  $GA_{20}$  glucosides (1c, 1d, 3c, 7a) elute prior to the glucosyl esters (1b, 3b, 6a), whereas the glucosyl conjugates of  $GA_4$  (2b, 2c) and  $GA_7$  (4b, 4c) show the opposite pattern.

Gibberellin  $A_{20}$ -13-0-glucoside (7*a*) and  $GA_{20}$  glucosyl ester (6*a*) are likely to occur endogenously. Their presence has been assumed in polar fractions after feeding  $GA_{20}$  to various plants (Frydman and MacMillan 1975, Durley et al. 1975, Yamane et al. 1975, 1977, Takahashi et al. 1976, Rood et al. 1982).

The significance of the bioactivities of GA conjugates is difficult to assess, since the observed responses may be due to hydrolysis of the sugar moiety and liberation of "free" gibberellin. For this reason it has been stated that GA conjugates per se are biologically inactive (Sembdner et al. 1980, Schneider 1983). There are some examples where the observed bioactivities apparently parallel the percentage of free GA released (Hiraga et al. 1974, Liebisch 1974).

#### **GA** Glucosyl Derivatives

	Rt	K'	N
GA <sub>1</sub> (1a)	3.82	0.75	3,300
$GA_1$ glucosyl ether $(Ib)^a$	3.50	0.60	1,830
$GA_1$ -3-0-glucoside $(1c)^b$	3.23	0.48	1,500
$GA_1$ -13-0-glucoside (1d) <sup>b</sup>	3.13	0.44	1,000
$GA_4(2a)$	14.45	5.63	3,340
$GA_4$ glucosyl ester $(2b)^a$	7.42	2.40	1,640
$GA_4$ -3-0-glucoside (2c) <sup>b</sup>	10.97	4.03	2,770
$GA_5(3a)$	6.20	1.84	2,160
$GA_5$ glucosyl ester $(3b)^c$	4.58	1.10	2,500
$GA_5$ -13-0-glucoside $(3c)^b$	4.18	0.89	2,460
$GA_7(4a)$	12.50	4.73	3,600
$GA_7$ glucosyl ester $(4b)^d$	6.25	1.87	2,750
$GA_7$ -3-0-glucoside (4c) <sup>e</sup>	10.03	3.60	2,520
$GA_{20}(5a)$	6.25	1.87	2,200
$GA_{20}^{-1}$ glucosyl ester (6 <i>a</i> )	5.17	1.37	1,960
$GA_{20}$ -13-0-glucoside (7 <i>a</i> )	4.25	0.95	2,600

**Table 1.** Retention times (Rt), capacity factors (K'), and number of theoretical plates (N) of GA glucosyl conjugates and their parent GAs on reverse-phase chromatography (Si Polyol C<sub>18</sub>), 1 ml min<sup>-1</sup>; methanol: 0.1% phosphoric acid = 55:45 (v:v).

<sup>a</sup> Synthesized according to Hiraga et al. (1974).

<sup>b</sup> Schneider et al. (1977b).

<sup>c</sup> Schneider et al. (1977a).

<sup>d</sup> Schneider et al. (1984).

e Schneider (1981b).

For this reason the activity of a GA conjugate should be compared only to the activity of the parent GA. In this approach the aglycone (GA) activity would be set at 100%, and the relative activities calculated from log-normal dose-response curves of both the aglycone and the conjugate. Comparison should be made of straight and parallel segments of the curves (Fig. 1).

Table 2 shows that all the GAs tested (GA<sub>1</sub>, GA<sub>4</sub>, GA<sub>5</sub>, GA<sub>7</sub>, GA<sub>20</sub>) are highly active in the *dwarf*-5 assay. The relative activities of all investigated glucosides of these GAs ranged from 1% to 5%. Assuming that each conjugate is inactive per se, only small percentages of the applied conjugates were hydrolyzed to give free, biologically active GAs. Our results support those of earlier studies on the activity of GA glucosides in maize and other bioassays that use the leaf as the site of treatment (Yokota et al. 1971, Sembdner et al. 1976). There are also no striking differences in activity between GA-3-0-glucosides and GA-13-0-glucosides, although these types were found to differ in their susceptibility to cellulase (Schliemann and Schneider 1979, Schneider and Schliemann 1979).

Relative bioactivities of 1% to 100% have been reported for the GA glucosyl esters of GA<sub>1</sub>, GA<sub>3</sub>, GA<sub>4</sub>, GA<sub>37</sub> (Hiraga et al. 1974). In our experiments the highest relative activity (50%) was found for GA<sub>4</sub> glucosyl ester (2b) (Table 2). All other investigated glucosyl esters were less active (10–20%) than (2b) but clearly more active than the corresponding glucosides. This could mean that



Fig. 2. Chemical structures of compounds referred to in the text.

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							Relative activity
	10 µg	5 µg	l µg	0.5 µg	0.1 µg	Control	(%)
GA <sub>1</sub> ( <i>Ia</i> )	128 ± 17	127 ± 10	103 ± 15	88 ± 12	$62 \pm 3$	45 ± 4	100
GA, glucosyl ester (1b)	$90 \pm 10$	$72 \pm 10$	61 ± 6	52 ± 4	$43 \pm 3$	$41 \pm 2$	10
GA <sub>1</sub> -3-0-glucoside (1c)	$50 \pm 4$	51 ± 7	$62 \pm 5$	54 ± 3	48 ± 3	$43 \pm 2$	$\overline{\nabla}$
GA <sub>1</sub> -13-0-glucoside (1d)	$70 \pm 7$	$60 \pm 6$	$61 \pm 6$	47 ± 6	1	41 ± 2	2-5
$GA_4$ (2a)	133 ± 16	$110 \pm 15$	88 ± 14	$66 \pm 10$	50 ± 4	$35 \pm 3$	100
GA4 glucosyl ester (2b)	$100 \pm 10$	$96 \pm 8$	$75 \pm 14$	64 ± 14	44 ± 4	$38 \pm 3$	50
$GA_4$ -3-0-glucoside (2c)	55 ± 6	54 ± 8	44 ± 3	41 ± 2	36 ± 4	38 ± 4	1-5
$GA_5(3a)$	$136 \pm 9$	$138 \pm 15$	$110 \pm 20$	$98 \pm 14$	69 ± 6	$37 \pm 3$	100
GA, glucosyl ester (3b)	$73 \pm 12$	$81 \pm 10$	$68 \pm 7$	55 ± 8	42 ± 6	$26 \pm 4$	5-10
$GA_{5}$ -13-0-glucoside (3c)	$61 \pm 6$	$54 \pm 10$	44 ± 7	38 ± 4	$32 \pm 3$	$27 \pm 3$	-
$GA_7 (4a)$	113 ± 8	$112 \pm 9$	96 ± 14	91 ± 16	76 ± 18	$17 \pm 2$	100
GA <sub>7</sub> glucosyl ester (4b)	$96 \pm 22$	$90 \pm 22$	$66 \pm 15$	$60 \pm 15$	$46 \pm 12$	$17 \pm 2$	10
$GA_7$ -3-0-glucoside (4c)	$52 \pm 11$	$47 \pm 12$	$36 \pm 5$	29 ± 7	19 ± 3	$17 \pm 2$	$\overline{\vee}$
$GA_{20}$ (5a)	128 ± 12	$120 \pm 10$	$109 \pm 10$	ļ	51 ± 6	$37 \pm 3$	100
GA <sub>20</sub> glucosyl ester (6a)	$104 \pm 11$	$101 \pm 16$	$74 \pm 6$	64 ± 5	50 ± 4	$40 \pm 3$	20
$GA_{20}$ -13-0-glucoside (7a)	$72 \pm 12$	$56 \pm 9$	$46 \pm 6$	$40 \pm 4$	I	35 ± 1	1–3

	10 <sup>-4</sup>	10-5	10-6	10 <sup>-7</sup> molar
GA <sub>1</sub> (1a)	$132 \pm 42 \\ 137 \pm 49 \\ 120 \pm 48 \\ 143 \pm 16$	$91 \pm 5$	$47 \pm 13$	$30 \pm 4$
GA <sub>1</sub> glucosyl ester (1b)		$93 \pm 27$	$43 \pm 12$	$32 \pm 7$
GA <sub>1</sub> -13-0-glucoside (1c)		$84 \pm 36$	$41 \pm 10$	$32 \pm 5$
GA <sub>1</sub> -13-0-glucoside (1d)		$77 \pm 29$	$40 \pm 2$	$28 \pm 5$
GA <sub>4</sub> (2a)	$125 \pm 17$	$46 \pm 5$	$42 \pm 4$	$24 \pm 4$
GA <sub>4</sub> glucosyl ester (2b)	$106 \pm 32$	53 ± 5	$36 \pm 3$	$30 \pm 7$
GA <sub>4</sub> -3-0-glucoside (2c)	$116 \pm 23$	50 ± 6	$34 \pm 4$	$22 \pm 3$
GA <sub>5</sub> (3a)	$117 \pm 29$	$81 \pm 35$	$77 \pm 38$	$42 \pm 8$
GA <sub>5</sub> glucosyl ester (3b)	$128 \pm 33$	76 ± 22	$65 \pm 22$	37 ± 7
GA <sub>5</sub> -13-0-glucoside (3c)	$122 \pm 20$	86 ± 21	$71 \pm 26$	36 ± 8
GA <sub>7</sub> (4a)	$122 \pm 23$	$65 \pm 7$	$40 \pm 4$	$29 \pm 6$
GA <sub>7</sub> glucosyl ester (4b)	$152 \pm 38$	$69 \pm 16$	$44 \pm 4$	$33 \pm 5$
GA <sub>7</sub> -3-0-glucoside (4c)	$141 \pm 28$	$72 \pm 26$	$48 \pm 8$	$32 \pm 6$
$GA_{20}$ (5a)	$113 \pm 14$	$87 \pm 27$	$51 \pm 14$	$36 \pm 4$
$GA_{20}$ glucosyl ester (6a)	$103 \pm 36$	$96 \pm 29$	$49 \pm 12$	$35 \pm 3$
$GA_{20}$ -13-0-glucoside (7a)	$110 \pm 16$	$90 \pm 23$	$52 \pm 9$	$29 \pm 6$
GA <sub>3</sub> Water control	140 ± 36	$133 \pm 28$ 29 ± 5	95 ± 12	39 ± 9

**Table 3.** Bioassay of some GA glucosides, GA glucosyl esters, and the parent GAs on dwarf rice cv. "*Tan-ginbozu*"; data are given as length (mm) of the seedling  $\pm$  confidence limit.

GA glucosyl esters, like their glucosides, are only partially hydrolyzed in the *dwarf*-5 assay to liberate free GAs (Liebisch 1974).

In the rice bioassay, where substances were applied via the roots, enzymes in the medium are apparently present that hydrolyze GA glucosyl conjugates. As a result high biological activities were observed for all compounds tested (Table 3). The dwarf rice assay is thus useful for the initial localization of GA glucosyl conjugates in crude or partially purified extracts.

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